

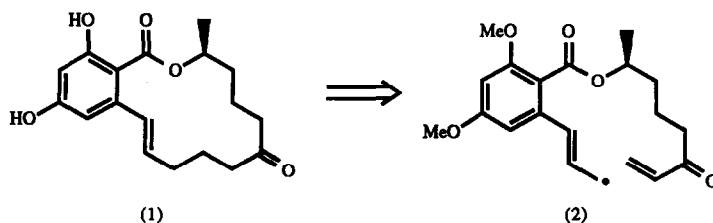
**SYNTHESIS OF MACROCYCLES VIA ALLYLIC RADICAL INTERMEDIATES.
A TOTAL SYNTHESIS OF (-)-ZEARALENONE**

Stephen A. Hitchcock and Gerald Pattenden*

Department of Chemistry, The University, Nottingham NG7 2RD

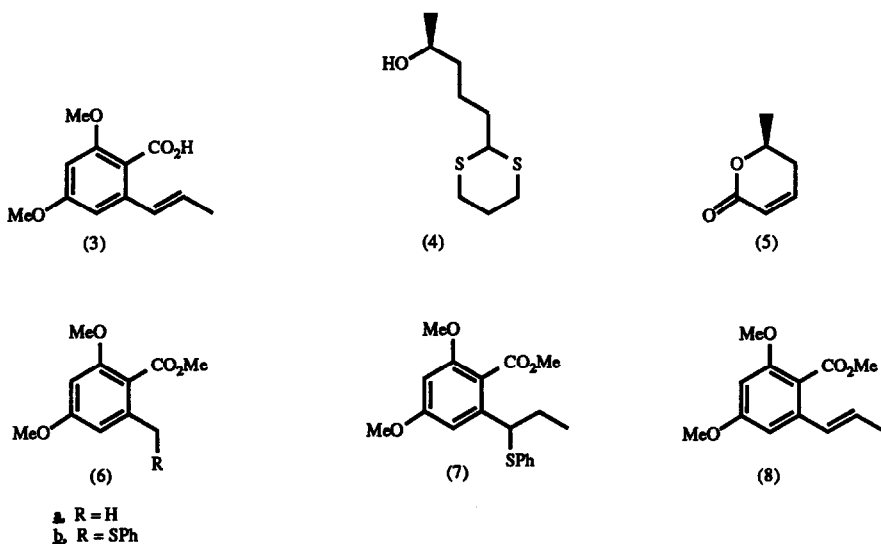
Summary: A concise synthesis of optically active (-)-zearalenone (1) which uses a novel 14-endo trig macrocyclisation from an allylic radical intermediate (Scheme 1) as a key feature, is described.

The 14-membered macrolide zearalenone (1) is an oestrogenic mycotoxin produced by various fusaria which colonise maize, barley, oats and wheat.¹ First isolated from the mycelium of the fungus *Gibberella zeae* (*Fusarium graminearum*), zearalenone is just one member of a growing family of biologically important "resorcyclic acid lactones" (RAL's) which have been found in nature.² Although a number of syntheses of racemic zearalenone have been published,³ to our knowledge no total synthesis of natural \underline{S} -zearalenone has been described. In earlier work we have described the use of allylic radical intermediates in macrocyclisation reactions leading to members of the cembranoid family of natural diterpenes.⁴ The presence of a δ -unsaturated ketone residue in zearalenone (1) led us to design a new synthetic strategy to the molecule based on the 14-endo-trig cyclisation from the cinnamyl radical intermediate (2) shown in Scheme 1. In this Letter we describe the successful outcome of this idea using the resorcinol derivative (3), and the chiral alcohol (4) derived from naturally occurring parasorbic acid (5), as key intermediates.



Scheme 1

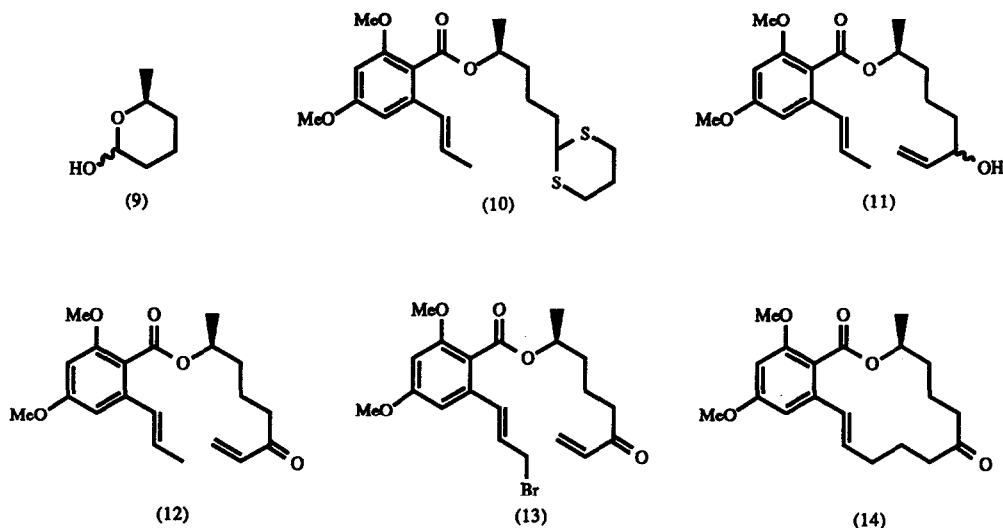
Thus, deprotonation of the readily available methyl orsellinate derivative ($6a^5$) (LDA, THF, -78°C), followed by sulphenylation of the resulting carbanion using diphenyl disulphide, first led to the phenyl sulphide ($6b$).⁶ Treatment of ($6b$) with potassium hexamethyldisilazide (THF, -78°C) followed by quenching with iodoethane next produced the substituted sulphide (7; 94%) which on oxidation (NaIO_4 , MeOH) and thermal elimination ($\text{C}_6\text{H}_5\text{Me}$, Δ , 2h) of phenylsulphenic acid gave rise to the *E*-alkene (8).⁷ Saponification of (8), using KOH-DMSO, then provided the resorcinol derivative (3), m.p. $85-6^\circ\text{C}$ (Et_2O -hexane); δ_{H} 1.9 (dd, J 6.7 and 1.6 Hz; $:\text{CHMe}$), 3.89 (OMe), 3.92 (OMe), 6.2 (dq, J 15.5 and 6.7 Hz, ArCH:CH), 6.4 (d, J 2.2 Hz, $:\text{CH}$), 6.5 (d, J 2.2 Hz, $:\text{CH}$), 6.8 (dq, J 15.5 and 1.6 Hz, ArCH:CH) p.p.m.



Sequential reduction (i.e. H_2 -Pd/C; then LiAlH_4) of naturally derived parasorbic acid (5)⁸ led to the cyclic hemiacetal (9) which was converted to the known dithiane (4)⁹ with propanedithiol in the presence of boron trifluoride. Treatment of the carbinol (4) with the acid chloride derived from (3) [$(\text{COCl})_2$, DMF, THF, 25°C] next led to the ester (10) as an oil, $[\alpha]_{\text{D}} + 17.6^\circ$ (c. 1.0, CHCl_3); ν_{max} (film) 1715 cm^{-1} ; δ_{H} 1.34 (d, J 6.8 Hz, CHMe), 5.2 (m, OCHMe) p.p.m.

Deprotection of (10), using HgO-HgCl_2 in aqueous acetonitrile then provided the corresponding aldehyde (54%) which upon treatment with vinylmagnesium bromide (THF, -78°C) led to the allylic alcohol (11; 78%). Oxidation of (11) using

manganese dioxide in dichloromethane finally gave the enone precursor (12), $[\alpha]_D + 27.2^\circ$ (c. 1.1, CHCl_3 ; ν_{max} (film) 1715 cm^{-1} ; δ_{H} 1.32 (d, J 6.8 Hz, CHMe), 4.9-5.4 (m, 2H), 5.8 (dd, J 9 and 4 Hz, $:\text{CH}$), 6.0-6.9 (m, 5H) p.p.m. to the radical intermediate shown in Scheme 1.



Controlled reaction between (12) and *N*-bromosuccinimide in the presence of ultraviolet light, produced exclusively the *E*-cinnamyl bromide (13) in a satisfying 62% yield.¹⁰ Finally, treatment of the bromide (13) with *tris*(trimethylsilyl)silane ($\text{C}_6\text{H}_5\text{Me}$, AIBN, 80°C , syringe pump over 8h)¹¹ resulted in clean 14-*endo*-trig macrocyclisation producing the known dimethyl ether (14) of (-)-zearalenone in 55% yield. Deprotection of the dimethyl ether (14)^{3f} then gave (-)-zearalenone (1), white crystals, m.p. $164-5^\circ\text{C}$ (Et_2O -hexane), $[\alpha]_D -191^\circ$ (c. 0.5, CHCl_3) which was identical in all respects (m.p. and mixed, optical rotation, chromatography, i.r., n.m.r.) with naturally derived material.

Acknowledgements

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References

1. For a recent review see: V.Betina, 'Zearalenone and its Derivatives', in *Mycotoxins Chemical, Biological and Environmental Aspects*, Elsevier, **1989**.
2. see refs 1 and 5, and also S.Lai, Y.Shizuri, S.Yamamura, K.Kawai, Y.Terada and H.Furukawa, *Tetrahedron Lett.*, **1989**, *30*, 2241.
3. a) A.V.Rama Rao, M.N.Deshmukh, and G.V.M.Sharma, *Tetrahedron*, **1987**, *43*, 779; b) T.Takahashi, H.Ikeda and J.Tsuji, *Tetrahedron Lett.*, **1981**, *22*, 1363; c) T.Takahashi, T.Nagashima and J.Tsuji, *Chem. Lett.*, **1980**, 369; d) T.Takahashi, K.Kasuga, M.Takahashi, and J.Tsuji, *J.Am. Chem. Soc.*, **1979**, *101*, 5072; e) N.N.Girotra and N.L.Wendler, *J. Org. Chem.*, **1969**, *34*, 3192; f) I.Vlattas, I.T.Harrison, L.Tökés, J.H.Fried, and A.D.Cross, *J. Org. Chem.*, **1968**, *33*, 4176; g) D.Taub, N.N.Girotra, R.D.Hoffsommer, C.H.Kuo, H.L.Slates, S.Weber, and N.L.Wendler, *Tetrahedron*, **1968**, *24*, 2443.
4. N.J.Cox, S.D.Mills and G.Pattenden, *Tetrahedron Lett.*, **1989**, *30*, 621.
5. A.G.M.Barrett, T.M.Morris, and D.H.R.Barton, *J. Chem. Soc., Perkin Trans 1*, **1980**, 2272.
6. F.M.Hauser, R.P.Rhee, S.Prasanna, S.M.Weinreb, and J.H.Dodd, *Synthesis*, **1980**, 72.
7. Satisfactory spectroscopic data, together with microanalytical and/or mass spectrometry data, were obtained for all new compounds. Optical rotations were measured at 25°C.
8. We thank Professor L.Crombie of this department for generous supplies of natural parasorbic acid. see: L.Crombie and P.Firth, *J. Chem. Soc., (C)*, **1968**, 2852.
9. F.W.Lichtenthaler, F.D.Klingler and P.Jarglis, *Carbohydr. Res.*, **1984**, *132*, C1-C4.
10. a) W.Offermann and F.Vögtle, *Angew. Chem. Int. Ed. Engl.*, **1980**, *19*, 464; b) W.Offerman and F.Vögtle, *Synthesis*, **1977**, 272. (Aromatic ring bromination becomes a problem unless carefully controlled conditions are adhered to.)
11. B.Giese, B.Kopping and C.Chatgillaloglu, *Tetrahedron Lett.*, **1989**, *30*, 681. The same 14-endo trigonal cyclisation was effected in the presence of Bu₃SnH-AIBN (~60%), but it was not always possible to remove small traces of tin residues from the final product.

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